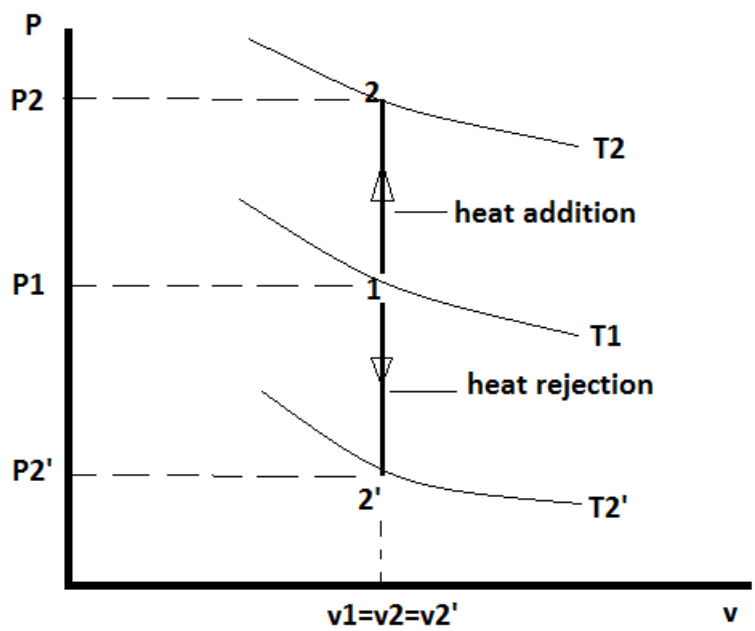


## NON-FLOW PROCESSES

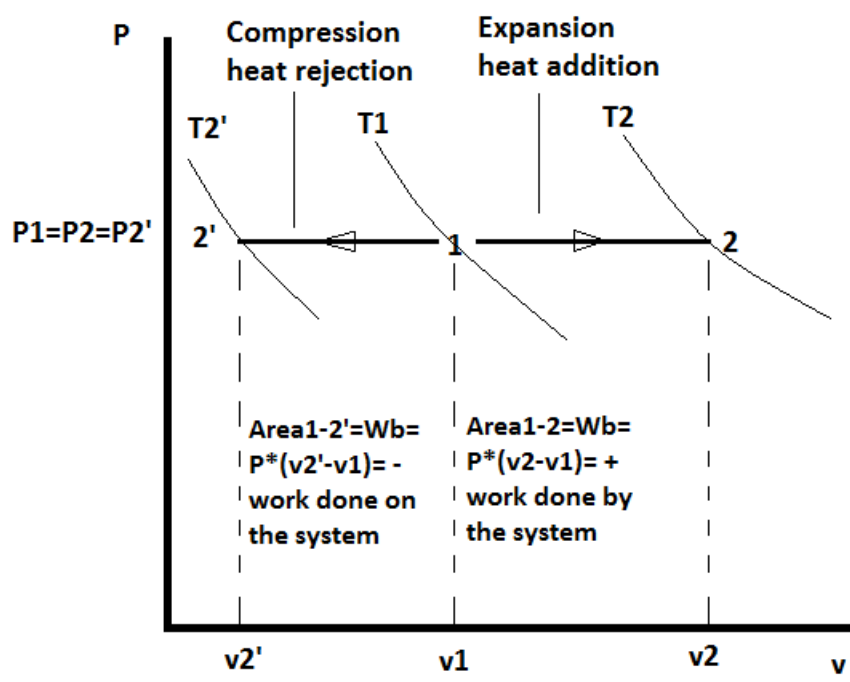
Working fluid: ideal gas

System: closed

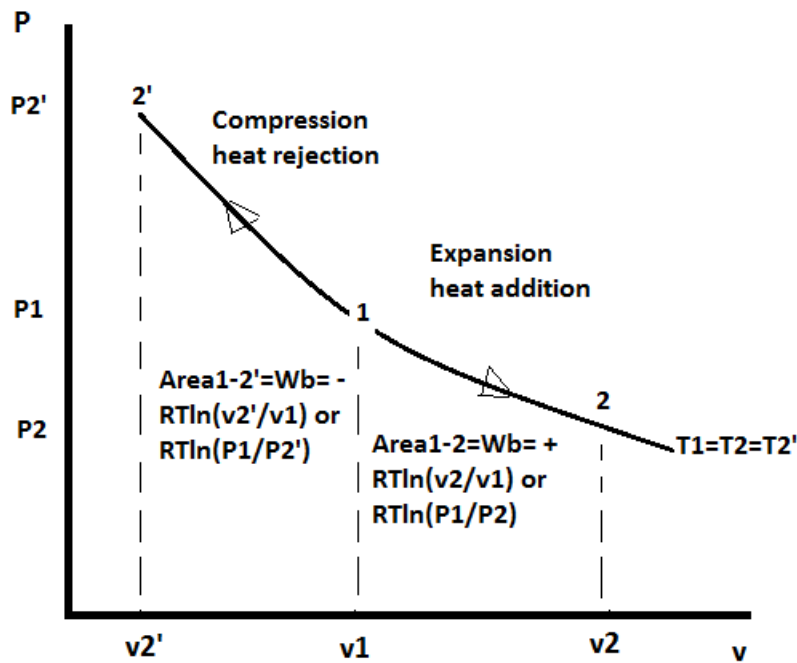
1. Process: Reversible constant volume



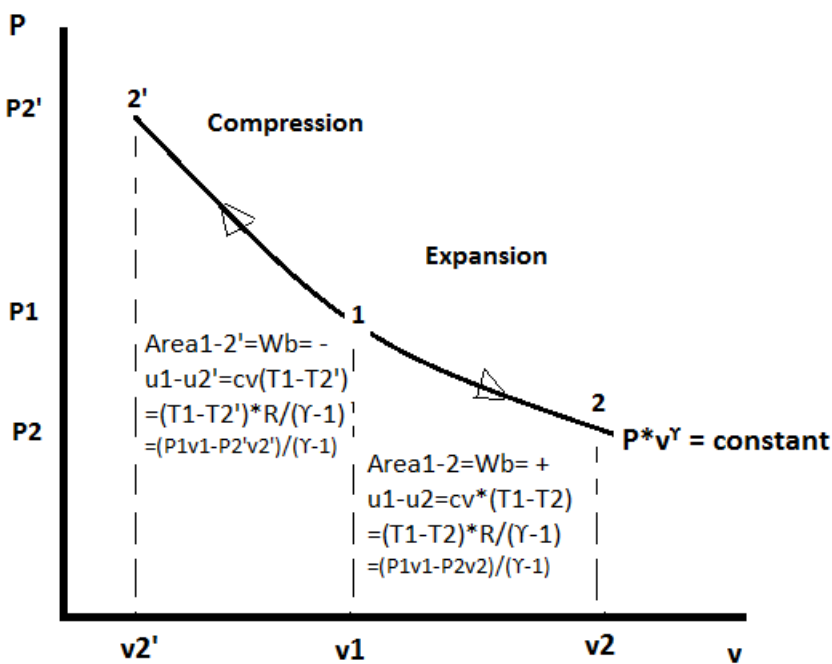
2. Process: Reversible constant pressure



Process: Reversible constant temperature (isothermal)

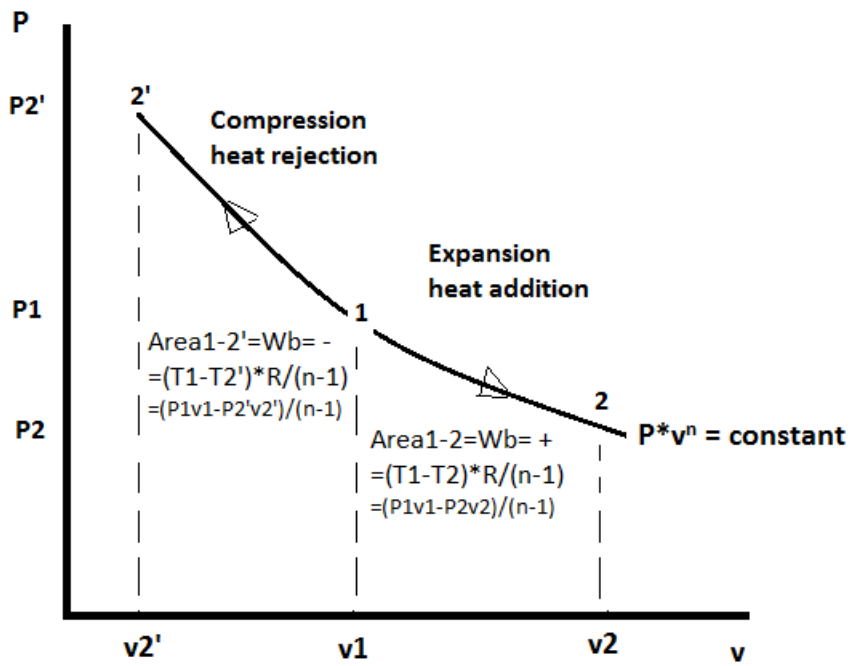


Process: Reversible adiabatic



$Q = 0$  and  $P, V, T$  change during reversible adiabatic process

Process: Polytropic



$$P \cdot v^n = \text{constant}$$

$$n \neq 0, 1, \gamma, \infty$$

P, V, T change during polytropic process